Claims 1-24 remain in this application with claims 1 and 13 in independent form.

Claims 1 and 13 have been amended. No new matter is believed to have been introduced

through these amendments.

Applicant has amended paragraphs [0010], [0026], and [0036] to correct

grammatical and typographical errors occurring therein. Specifically, in paragraph

[0010], "as" was inserted to correct a grammatical error, in paragraph [0026], "weight'

was inserted to correct a typographical error, and in paragraph [0036], the first occurrence

of "to" has been deleted. No new matter is believed to have been introduced through

these amendments.

Claims 1-24 stand rejected under 35 U.S.C. §102(b) as being anticipted by United

States Patent Nos. 5,489,366 to Jongenburger or 5,352,807 to Shih. The Examiner states

that both references disclose a purification method for purifying crude propylene oxide.

Applicant respectfully traverses the 35 U.S.C. §102(b) rejection on the grounds

that neither the '366 patent or the '807 patent discloses each and every element of the

claimed invention. Claims 1 and 13 have been amended to more particularly point out

and distinctly claim the subject invention.

Claim 1 is directed toward a method of purifying polyether polyols and claim 13

is directed toward a method of forming a polyurethane from the purified polyether polyol.

As discussed at length in the section titled "Background of the Invention" of the

specification as originally filed, numerous related art methods have been developed to

purify starting components, such as, ethylene oxide, propylene oxide, buytlene oxide, and

the like, which are used to form polyether polyols. However, it was experimentally

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discovered that polyurethane foams formed from these purified starting components

resulted in foams that were commercially unacceptable. It is believed that these results

occurred because impurities were present in the purified starting components.

As also discussed in the "Background of the Invention" section of the

specification as originally filed, it is believed that the shipping and storing of the purified

starting components allows impurities to become present and therefore, the purified

starting components are no longer pure and do not produce adequate foam.

Accordingly, the subject invention discovered that any additional impurities

present in the starting components could be removed after the polyether polyol has been

formed from the starting components. Claim 1, as amended, claims a method of

purifying polyether polyols of non-volatile impurities. The method comprises the steps of

providing a crude polyether polyol comprising a reaction product of an initiator and at

least one of propylene oxide and ethylene oxide.

The starting components for the polyether polyol include, but are not limited to,

propylene oxide, ethylene oxide, or mixtures of propylene oxide and ethylene oxide. As

understood by those skilled in the art, the polyether polyol may be all propylene oxide

with ethylene oxide end capping or vice versa. Alternatively, the polyether polyol could

be a mixture of ethylene oxide/propylene oxide and a mixture of ethylene

oxide/propylene oxide end caps.

Claim 1 has been amended to more particularly point out and distinctly claim that

the crude polyether polyol that is purified is formed from an initiator and a starting

component. The subject invention is not claiming a method of purifying the starting

components prior to forming the polyether polyol.

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After formation of the polyether polyol, at least one non-volatile impurity having

a number-average molecular weight of at least 25,000 Daltons is present in the polyether

polyol. It is believed that the impurity results from the storage or the shipping of the

starting components, such as the propylene oxide or the ethylene oxide, even after these

starting components have been purified. The crude polyether polyol is contacted with an

adsorbent for a time and under conditions sufficient to adsorb substantially all of the non-

volatile impurities onto the adsorbent and a purified polyether polyol is then separated

from the adsorbent.

Claim 13, as amended, claims a method of forming a polyurethane foam. The

method comprises the steps of contacting a crude polyether polyol with an adsorbent for a

time and under conditions sufficient to absorb substantially all of a non-volatile impurity

onto the absorbent. The crude polyether polyol comprises a reaction product of an

initiator and at least one of propylene oxide and ethylene oxide and has present therein

the non-volatile impurity. The non-volatile impurity has a number-average molecular

weight of at least 25,000 Daltons.

A purified polyether polyol is separated from the adsorbent after contacting the

crude polyether polyol with the adsorbent. The purified polyether polyol is then reacted

with at least one isocyanate to produce a polyurethane foam having increased stability

compared to a foam prepared using the crude polyether polyol.

Since neither the '366 patent or the '807 patent discloses, suggests, or teaches a

method of purifying a crude polyether polyol, claims 1 and 13 are believed to be

allowable. Claims 2-12 and 14-24, which depend directly or indirectly from claims 1 and

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13, are also believed to be allowable.

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Accordingly, it is respectfully submitted that the Application, as amended, is now

presented in condition for allowance, which allowance is respectfully solicited. Applicant

believes that no fees are due, however, if any become required, the Commissioner is

hereby authorized to charge any additional fees or credit any overpayments to Deposit

Account 08-2789.

Respectfully submitted

HOWARD & HOWARD ATTORNEYS, P.C.

January 25, 2005

Date

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CERTIFICATE OF MAILING

I hereby certify that this Amendment for United States Patent Application Serial Number 10/813,725 filed March 31, 2004 is being deposited with the United States Postal Service as First Class Mail, postage prepaid, in an envelope addressed to the Commissioner for Patents, P.O. Box 1450, Alexandria, Virginia 22313-1450 on <u>January</u> 25, 2005.

Melissa S. Dadisman

KKH/